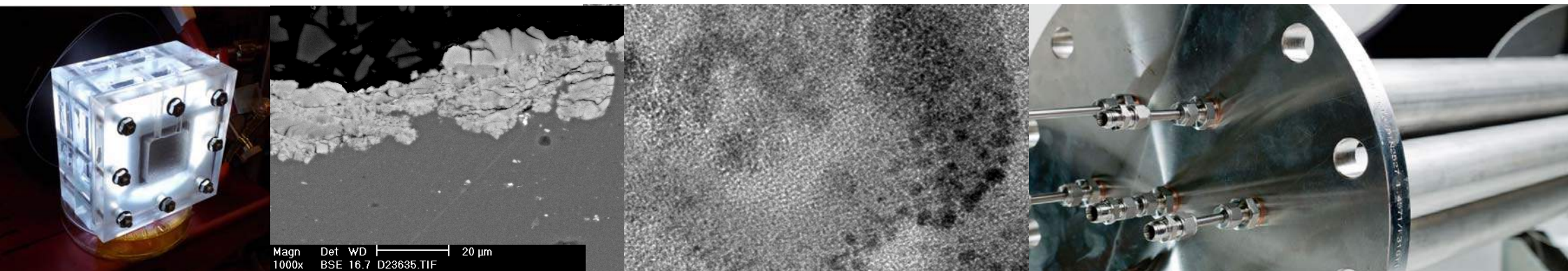
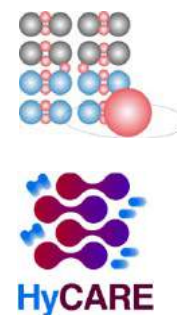


DAS HYCARE-PROJEKT – EINSATZMÖGLICHKEITEN VON METALLHYDRIDEN IN DER ENERGIESPEICHERUNG



Dr. Klaus Taube
Institut für Werkstofftechnologie - Nanotechnologie
HOST Speichertechnologien und Wasserstoff
Stralsund, 16.9.2019

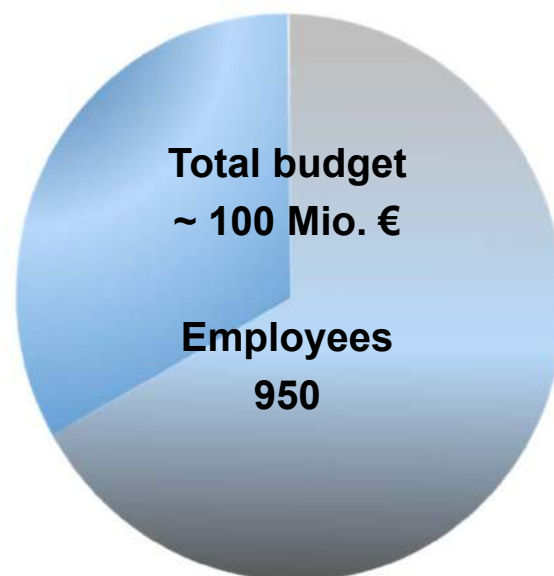


**Helmholtz-Zentrum
Geesthacht**
Centre for Materials and Coastal Research

HZG – PORTFOLIO

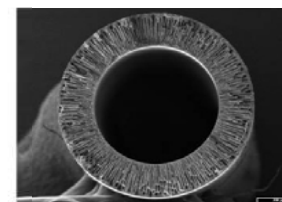
1/3

Coastal and Climate Research



2/3

Materials Research

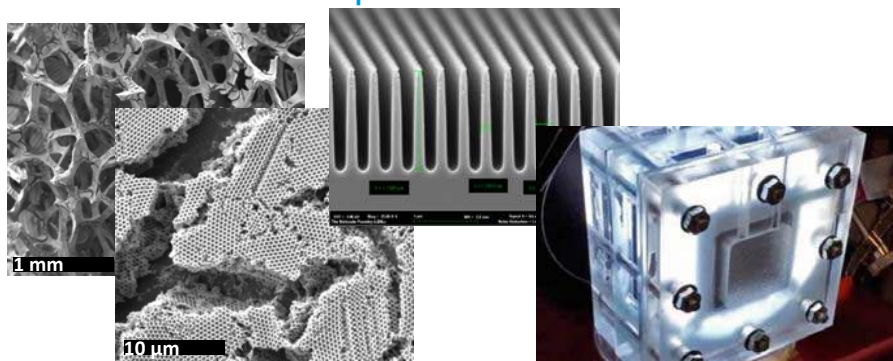


MATERIALS FOR HYDROGEN TECHNOLOGIES

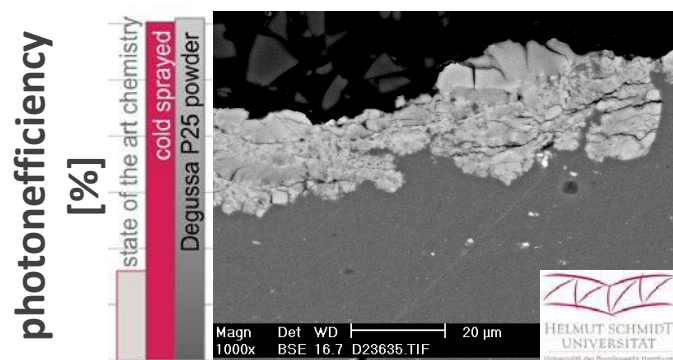
H₂ production by direct water splitting,
reversible storage from the gas phase in metal hydrides

Photocatalytic Films for H₂ Production

Hierarchically porous and nanoimprinting
structured photoelectrodes

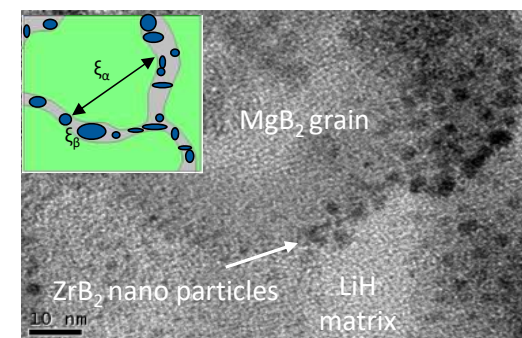


Electrodes prepared by cold-gas spraying

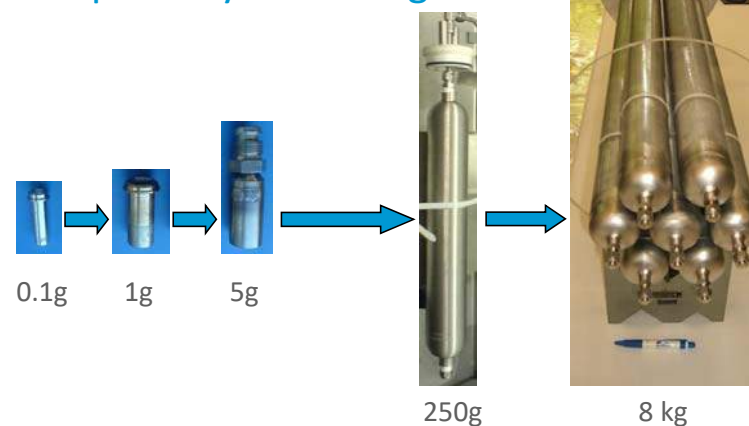


Materials and Systems for H₂ Storage

Optimised Hydrides with Low Mass & Volume



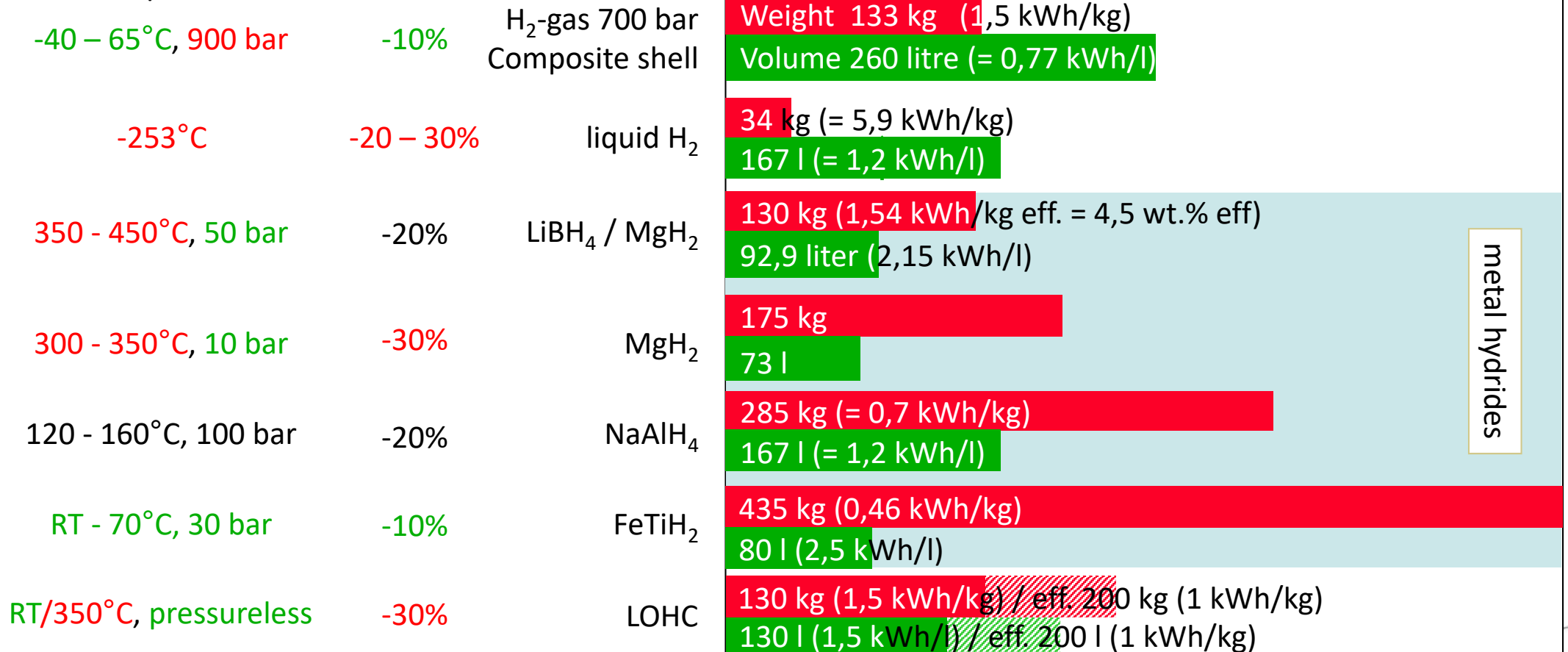
high performance tank design,
scale-up and system integration



COMPARISON OF STORAGE ALTERNATIVES

Tank weight and volume for 500 km range (6 kg H₂ = 200 kWh)

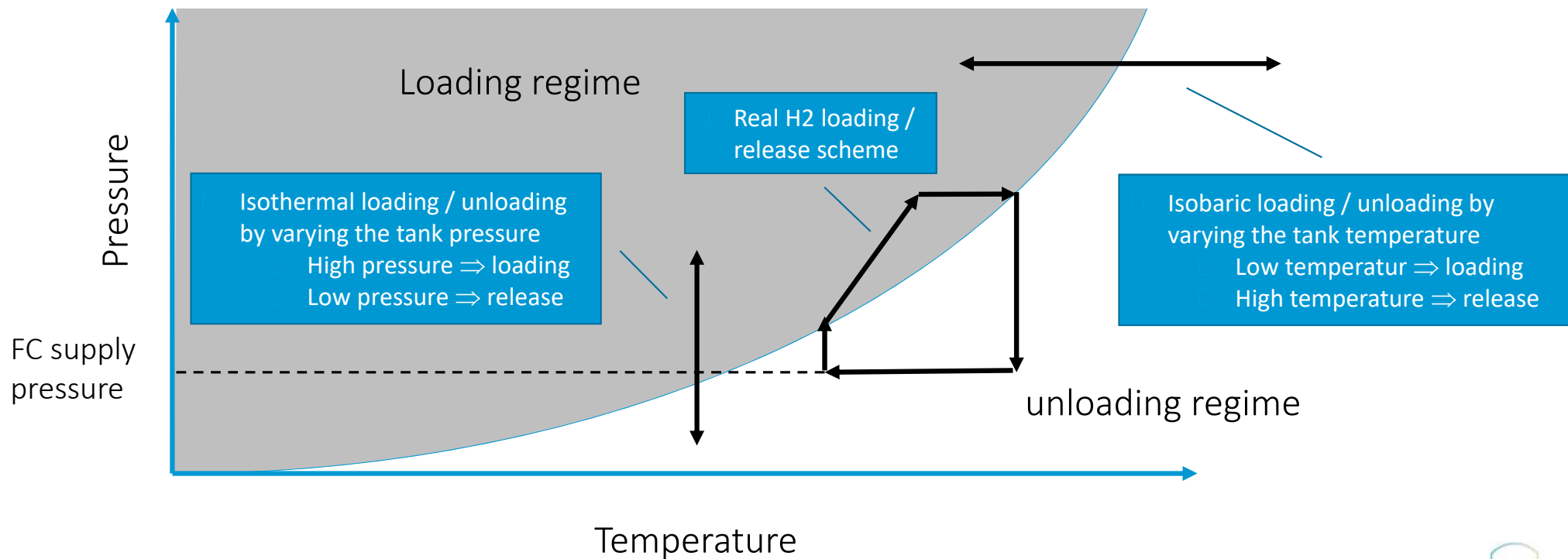
Temperature of operation, Efficiency
max. pressure



METAL HYDRIDES

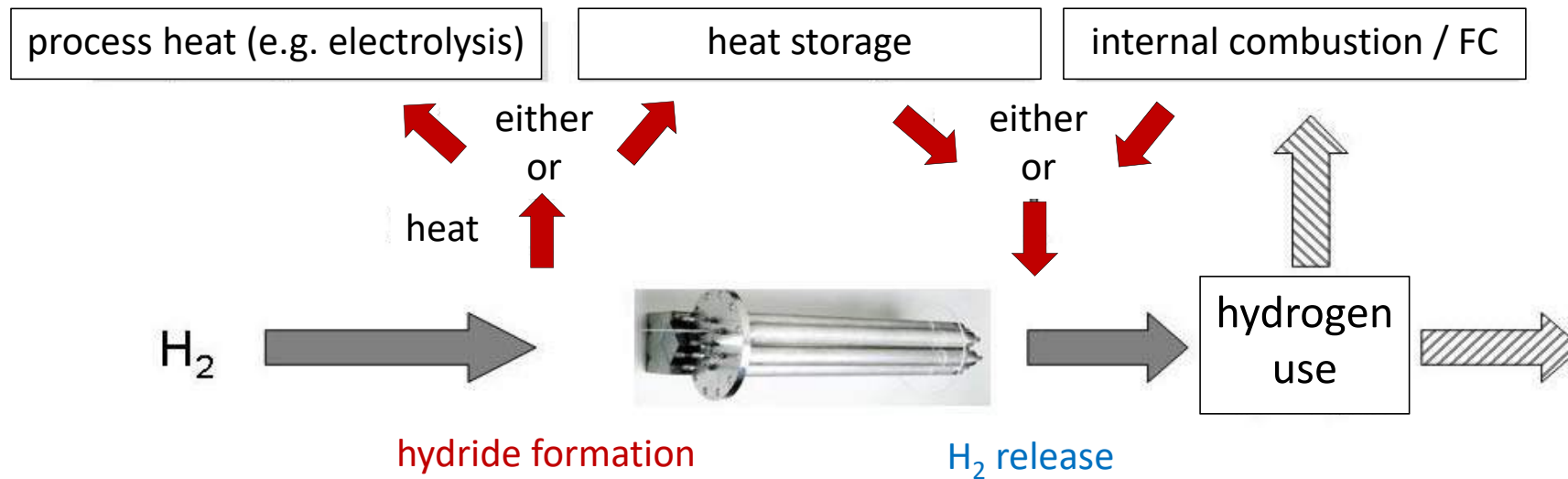
Principle of operation of a metal hydride store

Pressure / temperature equilibrium of the hydrogenation reaction



HEAT OF REACTION

high energy efficiency \Leftrightarrow heat management



- Application specific heat management necessary!!!

THE HYCARE PROJECT



FUEL CELLS AND HYDROGEN
JOINT UNDERTAKING



UNIVERSITÀ DEGLI STUDI DI TORINO



TECNO DELTA

Stühff GmbH 



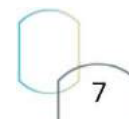
Helmholtz-Zentrum
Geesthacht
Zentrum für Material- und Küstenforschung



FONDAZIONE
BRUNO KESSLER



ENVIRONMENT
PARK
Parco Scientifico
Tecnologico per l'Ambiente

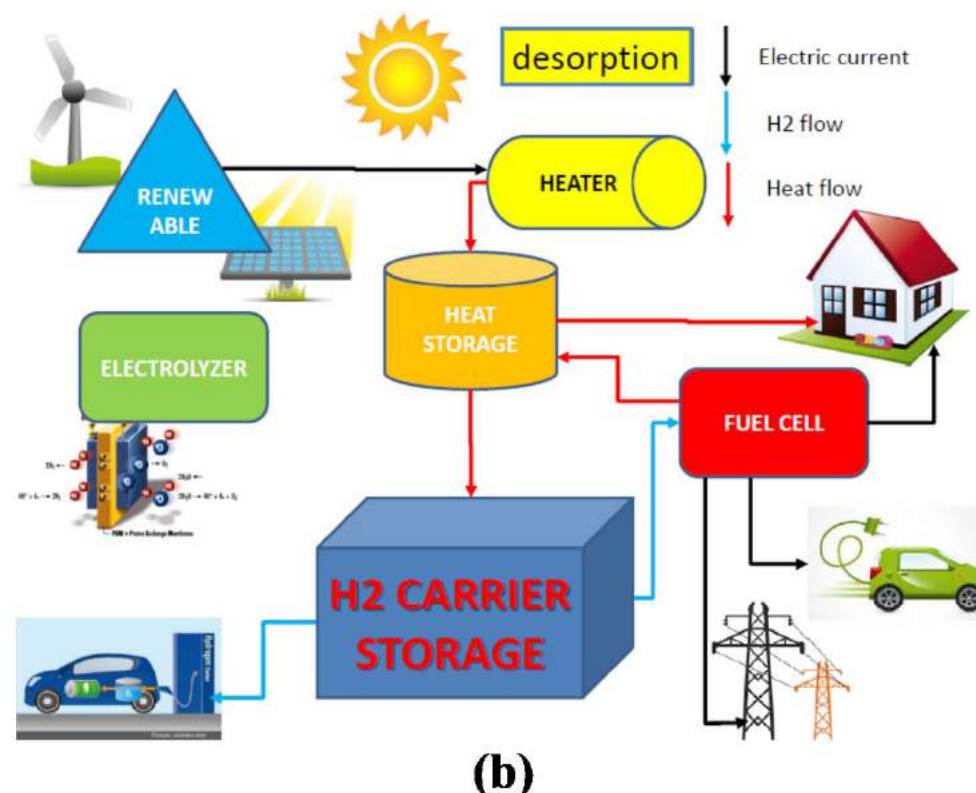
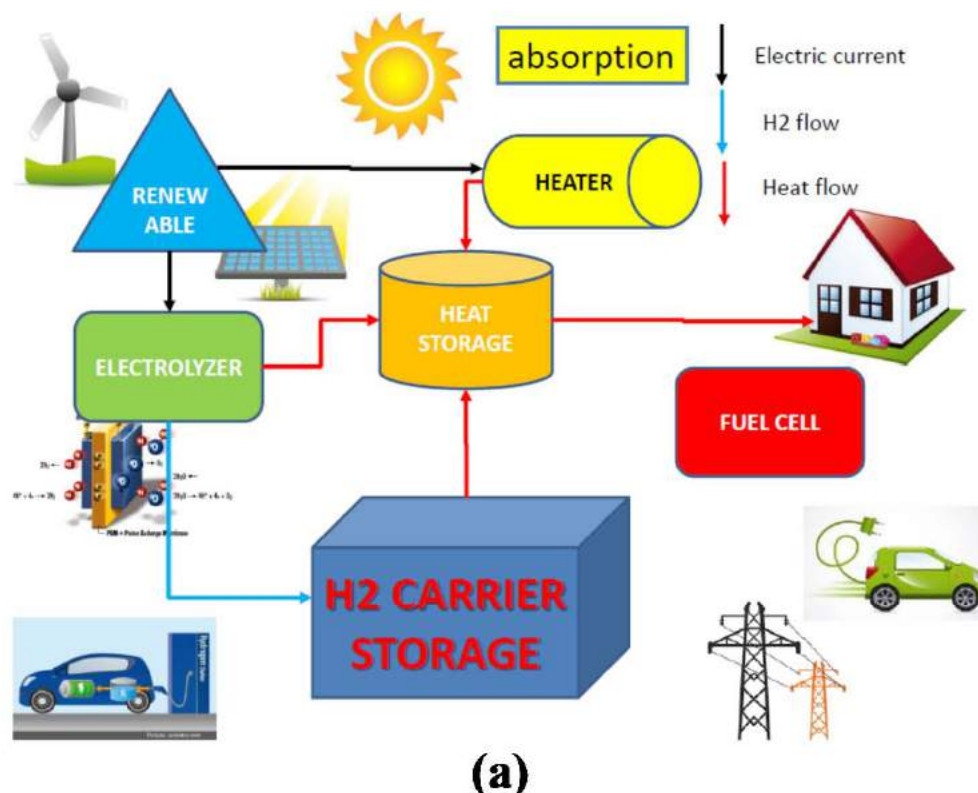


2019-2021, 2 Mio. €
FCH JU GA 826352

THE GOALS

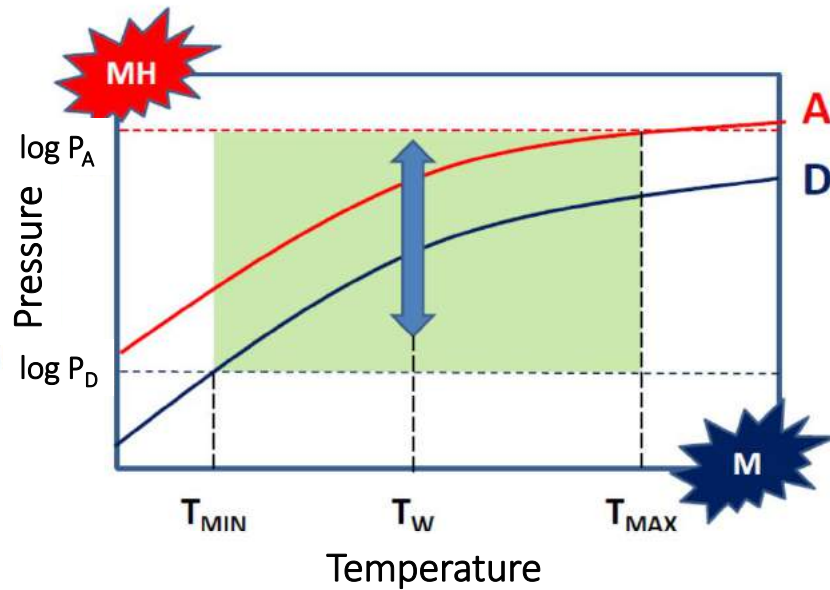
- High quantity of stored hydrogen ≥ 50 kg
- Low pressure < 50 bar, low temperature RT to $< 100^{\circ}\text{C}$
- Low foot print, comparable to liquid hydrogen storage (≥ 60 kg H_2/m^3)
- Hydrogen storage coupled with thermal energy storage \Rightarrow Improved energy efficiency
- Integration with an electrolyser (EL) and a fuel cell (FC) \Rightarrow Demonstration in real application
- Improved safety
- Techno-economical evaluation, Life Cycle Analysis (LCA)
- Exploitation of possible industrial applications
- Dissemination of results at various levels
- Engagement of local people and institution in the demonstration site

THE CONCEPT

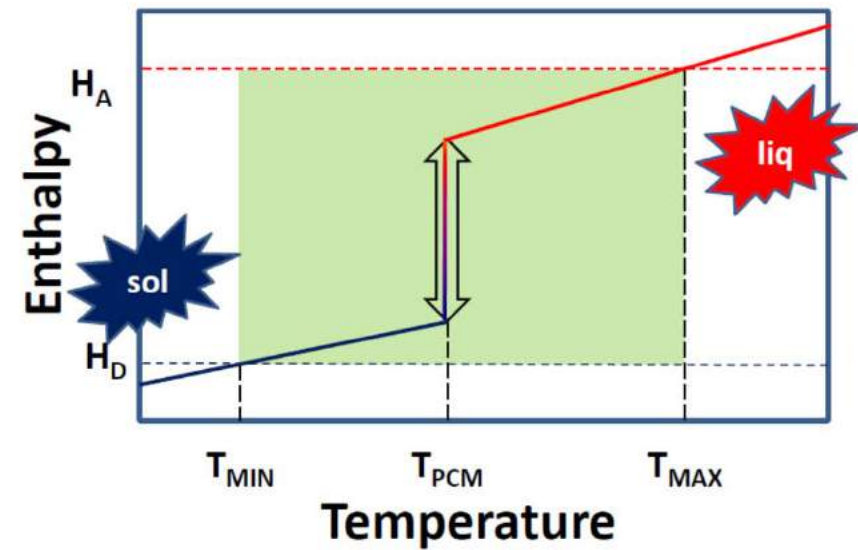


Flow of **HYDROGEN**, **HEAT** and **ELECTRICITY** during hydrogen production (a) and use (b)

H₂-CARRIER AND PCM

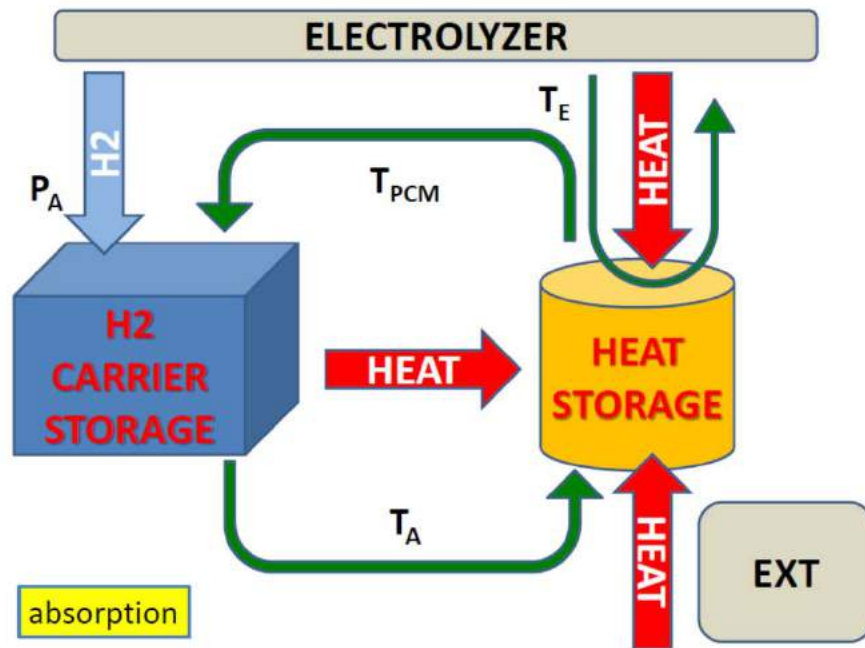


- P-T relationship of the hydrogen carrier during the absorption **A** and desorption **D** steps



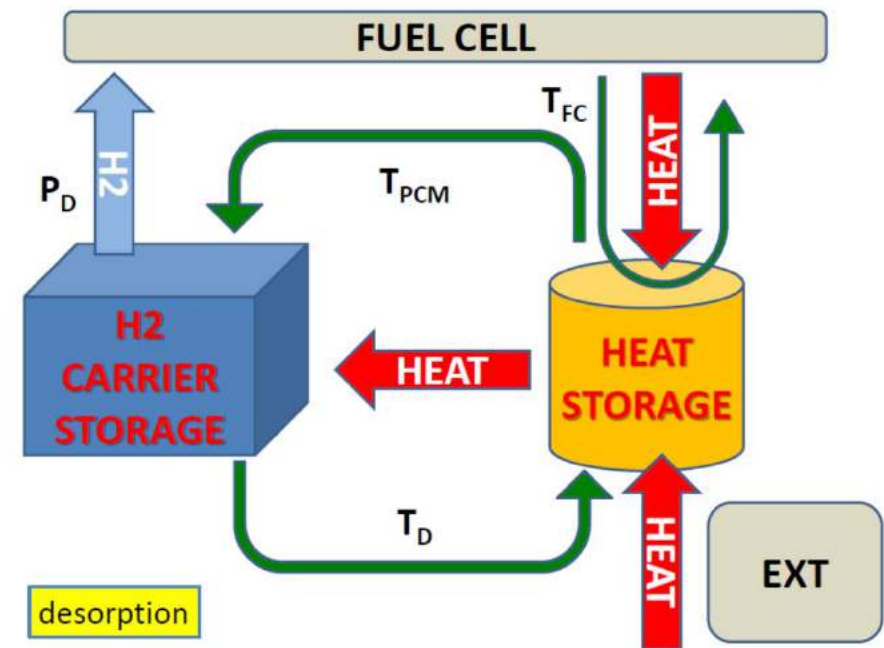
- E-T relationship for a phase change material during the absorption and desorption steps

THE INTEGRATION



(a)

- Thermal management during the hydrogen absorption step from the electrolyzer



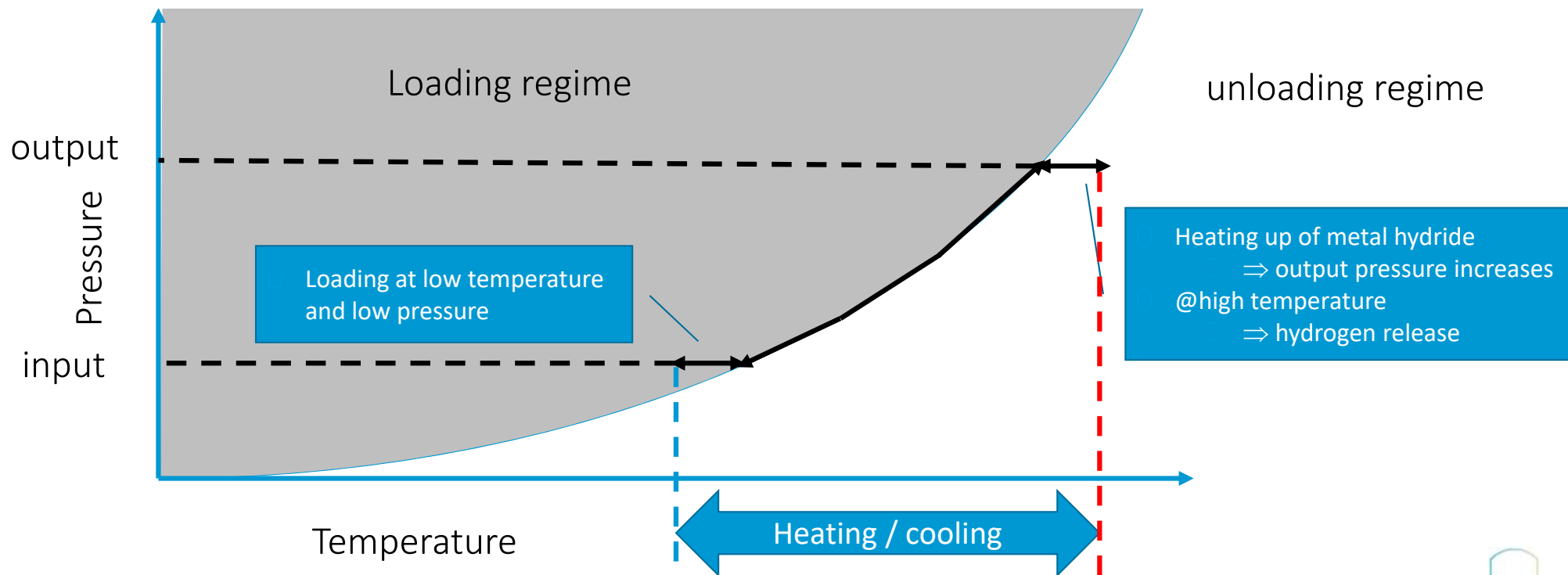
(b)

- Thermal management during the hydrogen desorption step to the fuel cell

METAL HYDRIDES

Operation of a metal hydride compressor

Pressure / temperature equilibrium of the hydrogenation reaction



METAL HYDRIDE COMPRESSORS

Compression without moving parts



200 bar MH compressor for refuelling of fork lifters
HYSA Systems, SA, 2015

M.V.Lototskyy, et al., Metal Hydride Hydrogen Storage and Supply
Systems for Electric Forklift with LT PEMFC Power Module, SDEWES2015, Dubrovnik,
Croatia, September 29, 2015

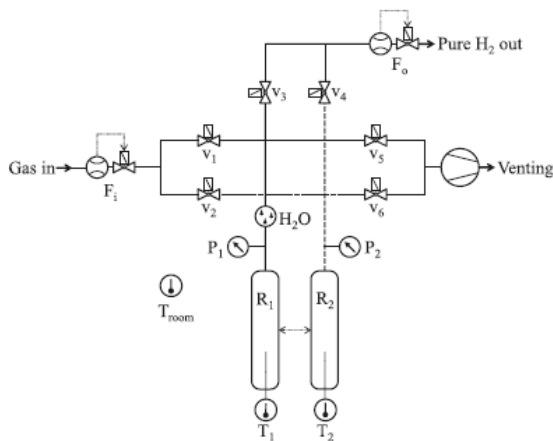


200 bar, 10 Nm³/h, smaller prototype integrated in
Lillestrom refuelling station, NO

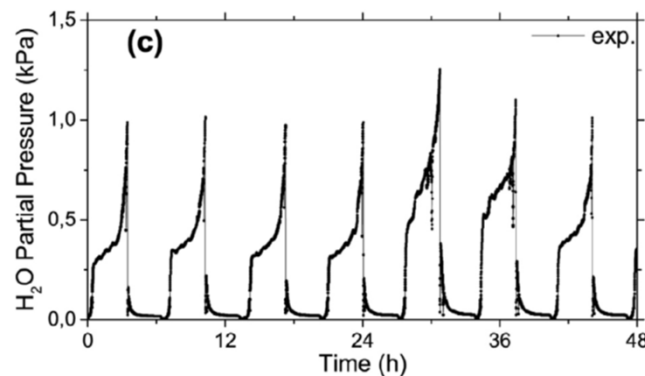
Hystorsys, 2013

<https://www.tu.no/artikler/denne-komprimerer-hydrogen-uten-annet-enn-ventiler-og-varme/276090>

HYDROGEN PURIFICATION



■ Basic reactor design (from [1])



■ Reduction of water level from 3000 ppmv to 190 ppmv (99,98%) @ 100 sccm H₂ flow (from [1])

■ Advantages:

- Just driven by gas pressure (high enough for MH loading!)
- No extra heating
- Selective reaction of the MH with H₂ ⇒ extraction from gas mixtures

■ Challenges:

- Poisoning of the MH (O, CO, ...)
- Optimal reactor design for required hydrogen flow
- Cost of the metal hydride
- Critical raw materials

- Ref. 4 (1987) in [2]: ammonia process exhaust (50.4 vol.% H₂, 25.5 vol.% N₂, 9.9 vol.% Ar, 12.4% CH₄, 1.8% NH₃) ⇒ 99.9% H₂
- Ref. 6 (1994) in [2]: NH₃ decomposition gas (50% H₂) with 12 Nm³/h and 24 Nm³/h ⇒ 99.9999% H₂

1. E.M. Borzone, A. Baruj, G.O. Meyer, Design and operation of a hydrogen purification prototype based on metallic hydrides, Journal of Alloys and Compounds 695 (2017) 2190 – 2198, <http://dx.doi.org/10.1016/j.jallcom.2016.11.067>
2. X. Chen, L. Wei, L. Deng, F. Yang, Z. Zhang, A review on the metal hydride based hydrogen purification and separation technology, Applied Mechanics and Materials, Vols. 448-453 (2013), pp 3027-3036, doi:10.4028/www.scientific.net/AMM.448-453.3027

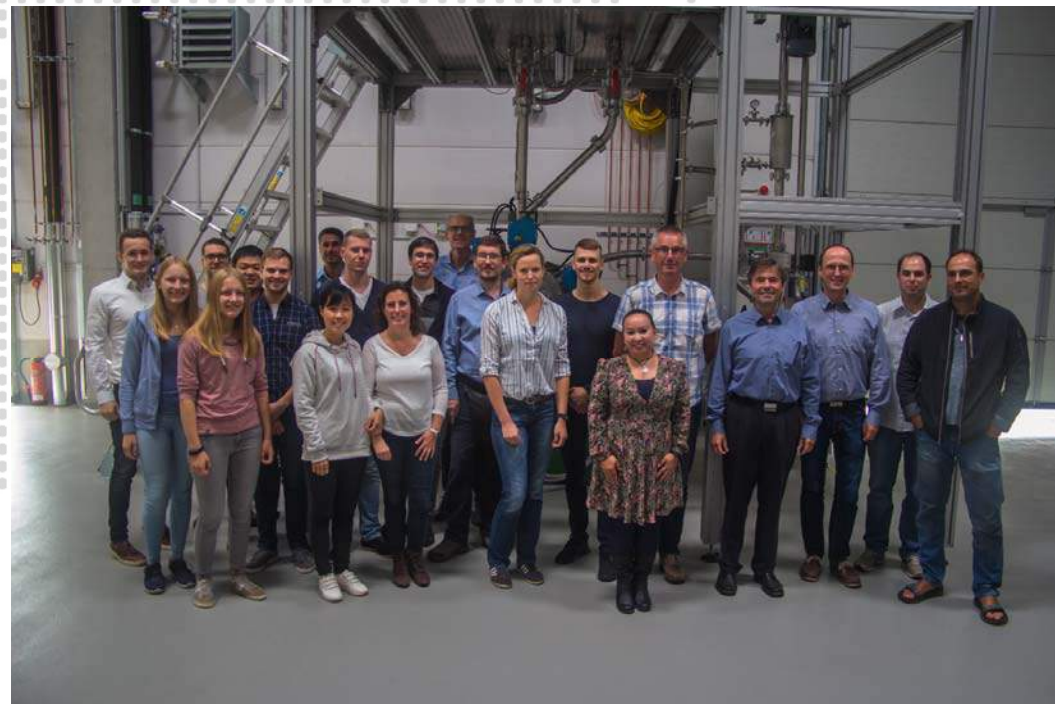
SUMMARY

Metal hydrides for hydrogen storage, compression and purification

- Lower volume than high pressure gas stores
- Direct loading with H₂ gas at pressures of 10 – 100 bar (depending on MH)
- No separate reactor for H₂ extraction
- No off-board regeneration necessary
- Low loading pressure ⇒ lower or no effort for compression
- Use as extra mass with energy storage function ⇒ forklifter, ships, mine locomotives
- Thermodynamic properties ⇒
 - Thermal compressors (supply from waste heat, minimised maintenance)
 - Hydrogen purification (practically no extra heat necessary, no compression)
- Challenge: materials cost – at present no large scale commercial production (except MH for NiMH batteries)
- Challenge: heat management simple to complex (depending on application and MH)

FOR PEOPLE AND THEIR FUTURE ENVIRONMENT

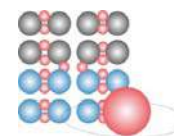
HyCARE has received funding from the Fuel Cells and Hydrogen 2 Joint Undertaking (JU) under grant agreement No 826352. The JU receives support from the European Union's Horizon 2020 research and innovation programme and Italy, France, Germany, Norway".



VIELEN DANK!

Dr. Klaus Taube
Institut für Werkstofftechnologie
klaus.taube@hzg.de
Tel. +49 (4152) 87 25 41

Max-Planck-Straße 1
22549 Geesthacht
wasserstoff.hzg.de



**Helmholtz-Zentrum
Geesthacht**
Centre for Materials and Coastal Research